

# Direct fermentation of Laminaria japonica for biohydrogen production by anaerobic mixed cultures

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#### ABSTRACT

A few studies have been made on fermentative hydrogen production from marine algae, despite of their advantages compared with other biomass substrates. In this study, fermentative hydrogen production from *Laminaria japonica* (one brown algae species) was investigated under mesophilic condition ( $35 \pm 1$  °C) without any pretreatment method. A feasibility test was first conducted through a series of batch cultivations, and 0.92 mol H<sub>2</sub>/mol hexose<sub>added</sub>, or 71.4 ml H<sub>2</sub>/g TS of hydrogen yield was achieved at a substrate concentration of 20 g COD/L (based on carbohydrate), initial pH of 7.5, and cultivation pH of 5.5. Continuous operation for a period of 80 days was then carried out using anaerobic sequencing batch reactor (ASBR) with a hydraulic retention time (HRT) of 6 days. After operation for approximately 30 days, a stable hydrogen yield of 0.79  $\pm$  0.03 mol H<sub>2</sub>/mol hexose<sub>added</sub> was obtained. To optimize bioenergy recovery from *L. japonica*, an up-flow anaerobic sludge blanket reactor (UASBr) was applied to treat hydrogen fermentation effluent (HFE) for methane production. A maximum methane yield of 309  $\pm$  12 ml CH<sub>4</sub>/g COD was achieved during the 90 days operation period, where the organic loading rate (OLR) was 3.5 g COD/L/d.

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# 1. Introduction

Energy supply and environmental protection are two crucial issues for the sustainable development of global prosperity. Over 80% of the energy consumed today in the world is derived from fossil fuels [1]. However, this current energy system is now facing two fundamental problems: gradual depletion and environmental pollution. This lack of sustainability has led researchers to search for new alternative energy sources [2]. Among various kinds of energy sources,  $H_2$  is regarded as the most promising future energy carrier, because it produces only water upon combustion, generating a higher energy yield (122 kJ/g), which is 2.75 times greater than that of hydrocarbon fuels. In addition, hydrogen can be

easily used as an automotive fuel in conventional internal combustion engines, and also can be applied in proton exchange membrane fuel cell vehicles [3].

 $H_2$  is commercially produced by either electrolytic or thermo-chemical processes, both of which are energy intensive [4]. From an environmental engineering point of view,  $H_2$  made from renewable resources seems to be more promising, since it meets the goal of sustainable development. In this regard, fermentative hydrogen production, where carbohydrates are directly fermented into  $H_2$ ,  $CO_2$ , and organic acids/alcohols without any external energy or electron acceptors, is considered a feasible biological process to produce  $H_2$  [5].

One of the main concerns in fermentative hydrogen production (FHP) is the high cost of the feedstock. In efforts to

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resolve this problem, many researchers have recently focused on lignocellulosic materials, which are composed of cellulose, hemicelluloses, and lignin, as new fermentative  $H_2$  production substrates [5–9].

Lignocellulosic biomass in nature is by far the most abundant raw material, originating from hardwood, softwood, grasses, and agricultural residues. The annual yields of lignocellulosic biomass residues worldwide are estimated to exceed 220 billion tons, equivalent to 60-80 billion tons of crude oil [10]. However, yields of H<sub>2</sub> produced by direct fermentation of lignocellulosic biomass are very low, mainly due to the complex structure of these substrates [11]. In order to enhance the digestibility of lignocellulosic material, different pretreatment methods have been applied, such as thermal, mechanical, acid, alkaline pretreatment, etc. However, various kinds of inhibitors are generated during these processes [6].

Marine algae is an aquatic group of cellulosic biomass. Although it has not been actively explored as an energy crop, it has many advantages for FHP, including the followings: 1) The main components of marine algae are cellulose and hemicellulose, not lignin, and thus fewer inhibitors will be generated during the pretreatment or fermentation process; 2) It has higher carbohydrate content compared with lignocellulosic biomass; and 3) It is massively abundant and easy to obtain or harvest [12,13]. Among this group, Laminaria japonica, a brown algae species, is a potential candidate for H<sub>2</sub> fermentation [14]. In 2006, the production of L. japonica in South Korea was estimated at around 0.8 million tons [14]. Like other brown algae, the main carbohydrate constituents of L. japonica are mannitol, laminaran, cellulose, and fucoidan and alginic acid, some of which are already known as good substrates for FHP [15,16]. Moreover, Jung et al. [16] reported that among various marine algae candidates, L. japonica showed the highest potential for FHP.

The aim of this study was to establish a stable FHP system using *L. japonica* as a feedstock. After a feasibility test under a series of batch cultivations, continuous operation was conducted using an anaerobic sequencing batch reactor (ASBR). To optimize bioenergy recovery, an up-flow anaerobic sludge blanket reactor (UASBr) was applied to treat the hydrogen fermentation effluent (HFE) for methane production.

# 2. Materials and methods

#### 2.1. Seed sludge and substrate

The seed sludge was taken from an anaerobic digester in a local wastewater treatment plant in South Korea. The pH, alkalinity, and volatile suspended solid (VSS) concentration of the sludge were 7.6, 2.83 g CaCO<sub>3</sub>/L, and 5.5 g/L, respectively.

For screening hydrogen producing bacteria (spore-forming anaerobic bacteria such as *Clostridium sp.*) and inactivating hydrogen consumers, 20 min heating at 90 °C was applied as a pretreatment step.

The feedstock was first dried at room temperature and then ground into 0.5 mm (diameter) particles by a normal blender. There was no external nutrient addition. The composition of *L. japonica* is shown in Table 1.

#### 2.2. Batch test

To investigate the feasibility of utilizing L. japonica for biohydrogen production and to determine optimal operation parameters, three batch tests were conducted under mesophilic condition (35  $\pm$  1 °C). Batch reactors with a working volume of 3 L were seeded with heat-pretreated sludge, equivalent to 30% of the working volume, and filled with a specific amount of L. japonica particles and tap water. The reactor was purged with N<sub>2</sub> for 5 min to provide an anaerobic condition and agitated at 150 rpm pH was monitored by pH sensors and controlled by the addition of 3N KOH solution. The produced gas was collected by gas collectors and sampled using a 1 ml syringe to analyze H<sub>2</sub> content. Three operation parameters, substrate concentration, initial pH, and cultivation pH (initial pH was controlled at the beginning of experiment, to provide suitable growth environment for H<sub>2</sub> producing microbes, and cultivation pH was controlled as constant during the fermentation process to ensure the microbial metabolic pathway was suitable for H<sub>2</sub> production and also inhibit H<sub>2</sub>consuming methanogenic activity), were evaluated in terms of their effect on hydrogen production. In the first batch test, the substrate concentrations were 5, 10, 20, 30, and 40 g Carbo. COD/L (calculation based on carbohydrate content and TCOD/TS ratio of the substrate); in the second batch test, the initial pH values were 7.0, 7.5, 8.0, 8.5, and 9.0; and in the third batch test, the cultivation pH values were 5.0, 5.5, 6.0, and 6.5, respectively. In the first batch test, the initial pH and cultivation pH values were kept at 8.0 and 5.5, while for the second and third batch tests, the operation parameters were selected as the optimal values based on the previous batch results (as shown in the result part, for the second batch test, the substrate concentration was 20 g Carbo. COD/L, and cultivation pH was 5.5; for the third batch test, the substrate concentration was 20 g Carbo. COD/L, and initial pH was 7.5).

To describe the hydrogen production, cumulative  $H_2$  production curves were obtained using the modified Gompertz Eq. (1) [17].

$$H(t) = P \times \exp\left\{-\exp\left[\frac{R \times e}{P}(\lambda - t) + 1\right]\right\}$$
(1)

Table 1 — Composition of Laminaria japonica.								
Name Composition (%)								
Laminaria	Carbohydrate					Protein	Lipid	Etc.
Japonica	Total	Cellulose	Hemi- cellulose	Lignin	Etc.			
	56.4	16.9	31.0	0	8.5	8.4	1.6	33.6

where  $H(t) = \text{cumulative H}_2 \text{ production (L) at cultivation time t}$ (hr); P = ultimate H<sub>2</sub> production (L); R' = H<sub>2</sub> production rate (L/L/hr);  $\lambda = \text{lag phase (hr)}$ ; and  $e = \exp(1) = 2.71828$ .

## 2.3. ASBR and UASBr operation

A schematic diagram of ASBR and UASBr is presented in Fig. 1. In this study, an ASBR with 5 L working volume was seeded with 30% of heat-treated sludge and filled with the substrate at 35  $\pm$  1 °C. The anaerobic condition was provided by  $N_2$ purging, the mixing ratio was 150 rpm, and pH was controlled by the addition of 3N KOH solution. The substrate concentration, initial pH, and operational pH were 20 g Carbo. COD/L, 7.5 and 5.5 (determined by batch tests). Continuous operation was delayed until approximately 0.5 mol H<sub>2</sub>/mol hexose<sub>added</sub> of H<sub>2</sub> was produced in the batch operation [18]. For ASBR operation, the hydraulic retention time (HRT) was 6 days, and the batch cycle was operated at 36 h, where 0.5 h was for feeding, 32 h for reaction, 3 h for settling, and 0.5 h for decanting. Once stable H<sub>2</sub> production was achieved after 30 days of continuous operation, the hydrogen fermentation effluent (HFE) was collected and used as a substrate for methane production in an UASBr with a working volume of 3.5 L. The granular sludge was obtained from a brewery wastewater treatment plant. And as substrate, HFE was centrifuged to remove the solid particles. HRT was maintained at 2 days. In order to optimize methane production, the organic loading rate (OLR) was gradually increased from 1 to 5 g COD/L/d. The methane production rate and methane yield were measured daily.

#### 2.4. Analytical methods

 $H_2$  content in biogas was determined by a gas chromatography (GC, Cow Mac series 580) using a thermal conductivity detector and a 1.8 mm  $\times$  3.2 mm stainless-steel column packed with molecular sieve 5 A with  $N_2$  as a carrier gas. The contents of  $CH_4$ ,  $N_2$ , and  $CO_2$  were measured using a GC of the

same model noted previously with a 1.8 mm  $\times$  3.2 mm stainless-steel column packed with Porapak Q (80/100 mesh), using helium as a carrier gas. The temperatures of the injector, detector, and column were kept at 80, 90, and 50 °C, respectively, in both GCs. Volatile fatty acids (VFAs, C2–C6) and lactic acid were analyzed by high performance liquid chromatography (HPLC) (Finnigan Spectra SYSTEM LC, Thermo Electron Co.) with an ultraviolet (210 nm) detector (UV1000, Thermo Electron) and an 100 mm  $\times$  7.8 mm Fast Acid Analysis column (Bio-Rad Lab.) using 0.005 M H<sub>2</sub>SO<sub>4</sub> as a mobile phase. Carbohydrate was determined using the phenol-sulfuric acid method [19]. COD was measured according to Standard Methods [17]. RNA concentration was determined by the colorimetric method of Liwarska-Bizukoje E. and Ledakowicz [20].

#### 2.5. Microbial analysis

To analyze the microbial communities, DNA in the mixed sample during continuous operation was extracted using an Ultraclean Soil DNA Kit (Cat # 12800-50; Mo Bio Laboratory Inc., USA). The 16S rDNA fragments were amplified by polymerase chain reaction (PCR). The region corresponding to positions 357-518 in the 16S rDNA of Escherichia coli was PCR-amplified using the forward primer EUB357f (5<sup>'</sup>-CCTACGGGAGGCAGCAG-3<sup>'</sup>) with a GC clamp (5<sup>'</sup>-CGCCC-5' end to stabilize the melting behavior of the DNA fragments and the reverse primer UNIV518r (5'-ATTACC GCGGCTGCTGG-3'). The procedure for PCR-DGGE was described in a previous study by the authors [18]. After PCR amplification, PCR products were purified with using a Multiscreen Vacuum Manifold (MILLIPORE com., USA). All strands of the purified PCR products were sequenced with the primer EUB357f using an ABIPRISM Big Terminator Cycle Sequencing Kit (Applied Biosystems, USA) in accordance with the manufacturer's instructions. A search of GenBank was conducted using the BLAST program.



Fig. 1 - Schematic diagrams of (A) ASBR and (B) UASBr.

Table 2 – H <sub>2</sub> production and effluent compositions.									
Feasibility test		H <sub>2</sub> Production	H <sub>2</sub> yield (mol	H <sub>2</sub> production	RNA conc.ª	Organic acid (g COD/L)			
		(L)	H <sub>2</sub> /mol hexose <sub>added</sub> )	rate (L/L/h)	(mg/L)	Total <sup>b</sup>	HAc <sup>c</sup>	HPr <sup>d</sup>	HBu <sup>e</sup>
Substrate	5	1.4	0.55	0.06	110	4.8	3.6	0	1.2
Conc. (g Carbo.	10	3.5	0.69	0.10	124	8.4	5.3	0	3.1
COD/L)	20	9.4	0.92	0.26	138	19.1	13.6	1.6	3.9
	30	12.6	0.83	0.24	126	26.6	15.9	3.6	7.1
	40	11.6	0.58	0.16	116	26.7	4.7	12.7	20.7
Initial pH	7.0	8.8	0.86	0.15	137	13.2	9.1	11.7	3.0
	7.5	9.6	0.95	0.23	143	18.4	12.3	1.4	4.7
	8.0	9.2	0.91	0.21	142	17.4	10.6	1.5	5.4
	8.5	8.9	0.88	0.22	140	15.4	9.9	2.1	3.4
	9.0	7.8	0.77	0.14	129	12.1	5.7	1.6	4.8
Cultivation pH	5.0	8.7	0.86	0.14	131	15.7	6.3	1.1	8.3
	5.5	9.3	0.92	0.25	141	17.8	14.5	0.6	2.3
	6.0	9.3	0.91	0.23	139	17.7	14.2	1.1	2.4
	6.5	7.3	0.72	0.13	116	10.4	4.6	0	5.8

a RNA conc. = RNA concentration when  $H_2$  production finished.

b Total = sum of acetate, propionate, butyrate.

c HAc = acetate.

d HPr = propionate.

e HBu = butyrate.

# 3. Results and discussion

# 3.1. Feasibility test

 $H_2$  yield is generally a good indicator of the effectiveness of  $H_2$  production.  $H_2$  yield results obtained through three batch tests are shown in Table 2, and the maximum  $H_2$  yield of 0.92 mol  $H_2$ /mol hexose<sub>added</sub>, or 71.4 ml  $H_2$ /g TS in Table 3, was achieved under a substrate concentration of 20 g Carbo. COD/ L, initial pH of 7.5, and cultivation pH of 5.5. Compared with results from previous studies on  $H_2$  production from raw lignocellulosic biomass, this value is much higher, as shown in Table 3. Some studies have noted that laminarin, the second most abundant compound in brown algae, can be easily degraded by microbes, as laminarase is found in various kinds of microorganisms [21,22]. Thus, owing to high carbohydrate content, an absence of lignin, and high biodegradability, *L. japonica* is a highly feasible feedstock for FHP.

In terms of organic acid production during H<sub>2</sub> fermentation, acetate and butyrate are accompanied by H<sub>2</sub> production, while propionate is mainly produced in H<sub>2</sub> consuming reactions, while lactate is known as a byproduct that is not related to H<sub>2</sub> production. As provided in Table 2, lactate was not detected in any of the samples, while propionate was found in some of them, which implies that the heat treatment can effectively inactivate non-spore-forming and non-H2producing bacteria such as lactic acid bacteria (LAB). However, the propionic acid bacteria (PAB) were inhibited by heat-shock but not totally exterminated [18]. In previous studies, the butyrate/acetate (B/A) ratio has been used as an indicator for evaluating the effectiveness of biohydrogen production. Furthermore, Arooj et al. [23] reported that the relationship between B/A ratio and H<sub>2</sub> yield was linear. In contrast, in this study, there was no linear relationship between B/A ratio and

Table 3 – Comparison of H <sub>2</sub> production obtained in this study using Laminaria japonica with previous study using raw lignocellulosic biomass.							
Substrate	Temperature	Maximum hydrogen yield achieved	Reference				
Grass silage	70 °C	16 ml H <sub>2</sub> /g TVS	[9]				
Cornstalk	36 °C	3.16 ml H <sub>2</sub> /g TVS	[10]				
Cornstalk	36 °C	23.3 ml H <sub>2</sub> /g TVS	[24]				
Beer lees	36 °C	6.8 ml H <sub>2</sub> /g TVS	[25]				
Beer lees	35 °C	$3.79 \text{ ml H}_2/\text{g TS}$	[26]				
Laminaria janonica	35 °C	$71.4 \text{ m}H_{o}/\sigma TS$	In this study				



Fig. 2 – H<sub>2</sub> yield during ASBR operation.

Table 4 $-$ H $_2$ yield and effluent compositions at steady state.									
Reactor	H <sub>2</sub> yield (mol H <sub>2</sub> /mol hexose <sub>added</sub> )	Carbohydrate removal (%)		Organic acid (g/L)			Organic acid (g/L)		
			Total <sup>a</sup>	HAc <sup>b</sup>	HPr <sup>c</sup>	HBu <sup>d</sup>			
ASBR	0.79 ± 0.03	82.9 ± 1.7	13.9 ± 0.4	9.3 ± 0.3	1.0 ± 0.1	3.6 ± 0.3			
a Total = sum of acetate, propionate, butyrate. b HAc = acetate.									
2 HPr = propionate.									
d HBu = bu	. HBu = butyrate.								

Α B С D Ε E G Н

Fig. 3 – DGGE profile for bacterial diversity from mixed sample.

 $H_2$  yield, which indicates that the substrate type can strongly affect the microbial metabolic pathway.

VSS generally indicate the presence of microorganisms; however, VSS cannot differentiate organic solids and microorganisms. On the other hand, RNA, result from microbial metabolism, can reflect the microbial growth condition. Thus, in this study, RNA concentration from HFE was measured to show the active microbial concentration and thereby determine the effect of each operation parameter on H<sub>2</sub> production. From the results it was found that RNA concentration accorded with total organic acid production and also H<sub>2</sub> yield, as shown in Table 2. Both H<sub>2</sub> and organic acids are metabolic byproducts from the feedstock, generated through the growth of hydrogen producing microbes. It is therefore reasonable that the optimal growth condition would result in the highest microbial growth (shown as highest RNA concentration) and maximum generation of metabolic products. This is the first report describing the application of RNA concentration along with H<sub>2</sub> yield to suggest the optimal operation conditions for FHP.

## 3.2. Continuous operation

Fig. 2 shows the H<sub>2</sub> yield during ASBR operation. Initially, a drastic decrease of H<sub>2</sub> production was observed upon changing the operation mode from batch to continuous operation. Kim et al. [18] noted that this phenomenon might be due to the regrowth of PAB, which is inhibited by heat-shock but not totally exterminated. After continuous operation for about 30 days, a stable H<sub>2</sub> yield of 0.79  $\pm$  0.03 mol H<sub>2</sub>/ mol hexose<sub>added</sub> was obtained. In about 50 days, a steady state was achieved, as listed in Table 4. Acetate and butyrate were

Table 5 — Affiliation of DGGE fragments determined by their 16S rDNA and isolated microorganisms.						
Band	Affiliation	Similarity (%)	NCBI GenBank No.			
А	Clostridium sp.	92	DQ677024.1			
В	Ruminococcus sp.	85	AY669260.1			
С	Clostridium polysaccharolyticum	92	X77839.1			
D	Clostridium sp.	94	AY827856.1			
Е	Eubacterium limosum	89	U67159.1			
F	Clostridium sp.	96	DQ677005.1			
G	Uncultured Clostridiales bacterium	97	FR695912.1			
Н	Uncultured Clostridiales bacterium	96	HQ103933.1			
Ι	Uncultured Bacteroides sp.	88	HM754520.1			



the main VFA components while propionate was kept at low concentration and lactate was not detected. These results show that heat treatment of the inoculum was an effective means of inhibiting LAB and  $H_2$  consumers, and that the operation conditions and feedstock were favorable for  $H_2$  production.

In order to detect dominant microorganisms, a mixed sample was taken from ASBR and the bacterial diversity was

Table 6 – UASBr performance for CH <sub>4</sub> production.						
OLR (g- COD/L/d)	Methane yield (mL CH <sub>4</sub> /g COD)	Methane production rate (L CH4/d)	COD removal (%)			
1	$185\pm8$	$\textbf{0.65} \pm \textbf{0.03}$	95			
2	$223\pm11$	$1.56\pm0.08$	94			
2.5	$254\pm9$	$\textbf{2.23} \pm \textbf{0.08}$	94			
3	$279 \pm 10$	$\textbf{2.93} \pm \textbf{0.11}$	93			
3.5	$309\pm12$	$\textbf{3.80} \pm \textbf{0.15}$	94			
4	$293 \pm 14$	$\textbf{4.12} \pm \textbf{0.20}$	90			
5	$\textbf{273} \pm \textbf{16}$	$4.80\pm0.28$	86			

monitored by polymerase chain reaction-denaturing gradient gel electrophoresis (PCR-DGGE). From the DGGE profile (Fig. 3), a total of 9 DNA bands were detected, where each band represents one microbial species. The results of 16S rDNA sequences shown in Table 5 reveal that, among the total DNA bands, 6 matched well with Clostridium sp. (band A, C, D, F, G, H), while the remaining bands were matched to Ruminococcus sp., Eubacterium limosum, and Bacteroides sp., respectively. All of these detected bacteria are well known as H<sub>2</sub> producing bacteria and also cellulolytic bacteria [27-30]. Accordingly, high H<sub>2</sub> yield and high degradation efficiency (indicated by the carbohydrate removal rate) from this cellulosic biomass feedstock could be achieved even though no pretreatment method was applied. This is the first report documenting the existence of hydrogen producing bacteria in FHP from L. japonica.

#### 3.3. CH<sub>4</sub>-UASBr performance

Fig. 4 shows the results of UASBr operation for 90 days. Methane content in the biogas was around 70–74% during the whole operation period. From day 1 to day 50, as the OLR increased, both methane yield and methane production rate increased simultaneously. When the OLR was 3.5 g COD/L/d, maximum methane yield of  $309 \pm 12$  ml CH<sub>4</sub>/g COD was achieved, which was 88.3% of the theoretically achievable CH<sub>4</sub> yield. Further increase of the OLR caused a decrease of methane yield and COD removal efficiency, respectively, thus indicating that an OLR above 3.5 g COD/L/d was beyond the COD degradation capacity of the microorganisms in UASBr.

Table 6 shows the UASBr performance under different OLR. COD removal effective during the whole continuous operation period, indicating that the HFE is a favorable feedstock for methane production.

# 3.4. Bioenergy recovery efficiency and COD balance from organic solid particles

Bioenergy recovery and COD removal efficiency of the twostage fermentation system treating organic solids were evaluated if which the results were presented in Fig. 5. Optimal



Fig. 5 – Bioenergy recovery and COD removal efficiency of the two-stage fermentation system.

ASBR and UASBr performance (with highest  $H_2$  and  $CH_4$  yield) in this study were used in this evaluation.

It was found that 49.9% biogas conversion ( $H_2$  3.8%,  $CH_4$  46.1%) and 55.1% COD removal were achieved in this system. The overall performance was limited by UASBr, given that 41.8% COD remained as sediment by centrifugation. However, it was difficult to conclude whether UASBr was economically suitable for this two-stage system, because OLR and  $CH_4$  production rates were higher, and HRT was lower than that of other reactor types including CSTR and ASBR [31,32].

In addition to the change of reactor type for  $CH_4$  fermentation, further additional treatment on the sediment, such as post-treatment or anaerobic dry digestion process could be adopted to obtain more  $CH_4$  from the waste portion of the total bioenergy.

# 4. Conclusions

Direct fermentation of *L. japonica* for biohydrogen production was attempted through batch tests and continuous operation under mesophilic condition. The following conclusions have been drawn from this study:

- 1. In the feasibility test, high  $H_2$  yield of 0.92 mol  $H_2$ / mol hexose<sub>added</sub>, or 71.4 ml  $H_2$ /g TS, was obtained, indicating the strong potential of *L. japonica* for fermentative hydrogen production.
- 2. Stable H<sub>2</sub> production was achieved in the continuous operation for 80 days using ASBR, with a H<sub>2</sub> yield of 0.79  $\pm$  0.03 mol H<sub>2</sub>/mol hexose<sub>added</sub>. A microbial diversity analysis indicated that Clostridium sp. was the predominant bacterial group in this mixed cultures system.
- UASBr showed high performance to treat HFE. However, the total bioenergy recovery through this two-stage fermentation system was limited due to particulate solid portion that could not be fed to UASBr.

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#### REFERENCES

- Das D, Veziroglu TN. Hydrogen production by biological processes: a surveyof literature. Int J Hydrogen Energy 2001; 26:13–28.
- [2] Brey JJ, Brey R, Carazo AF, Contreras I, Hernandez-Diaz AG, Castro A. Designing a gradual transition to hydrogen economy in Spain. Int J Hydrogen Energy 2006;159:1231–40.
- [3] Momirlan M, Veziroglu TN. Current status of hydrogen energy. Renewable Sustainable Energy Rev 2002;6:141–79.
- [4] Rajeshwar K, Ibanez JG, Swain GM. Electrochemistry and the environment. J Appl Electrochem 1994;24:1077–91.

- [5] Hawkes FR, Hussy I, Kyazze G, Dinsdale R, Hawkes DL. Continuous dark fermentative hydrogen production by mesophilic microflora: principles and progress. Int J Hydrogen Energy 2007;32:172–84.
- [6] Hendriks ATWM, Zeeman G. Pretreatments to enhance digestibility of lignocellulosic biomass. Bioresour Technol 2009;100:10–8.
- [7] Fan Y, Xing Y, Ma H, Pan C, Hou H. Enhanced cellulosehydrogen production from corn stalk by lesser panda manure. Int J Hydrogen Energy 2008;33:6058–65.
- [8] Ahmad T, Masoumeh G Ahangari, Chirayu S, Ajay K Dalai. Production of hydrogen and syngas via gasification of the corn and wheat dry distiller grains (DDGS) in a fixed-bed micro reactor. Fuel Processing Technol 2009; 90:472–82.
- [9] Pakarinen O, Lehtomäki A, Rintala J. Batch dark fermentative hydrogen production from grass silage: the effect of inoculum, pH, temperature and VS ratio. Int J Hydrogen Energy 2008;33:594–601.
- [10] Zhang M, Fan Y, Xing Y, Pan C, Zhang G, Lay JJ. Enhanced biohydrogen production from cornstalk wastes with acidification pretreatment by mixed anaerobic cultures. Biomass Bioenergy 2007;31:250–4.
- [11] David B, Carlo R, Nazim C, Richard S. Challenges for biohydrogen production via direct lignocellulose fermentation. Int J Hydrogen Energy 2009;34:7390–403.
- [12] Tseng CK. Algal biotechnology industries and research activities in China. J Appl Phycol 2001;13:375–80.
- [13] Park JI, Lee JW, Sim SJ, Lee JH. Production of hydrogen from marine macro algae biomass using anaerobic sewage sludge microflora. Biotechnol Bioprocess Eng 2009;14:307–15.
- [14] Park JI, Wu HC, Lee JH. Production of bio-energy from marine algae: status and perspectives. Korean Chem Eng Res 2008; 46:833–44 [In Korean].
- [15] Elizabeth P. The polysaccharides of green, red and brown seaweeds: their basic structure, biosynthesis and function. Br Phycol J 1979;14:103–17.
- [16] Jung KW, Kim DH, Shin HS. Fermentative hydrogen production from Laminaria japonica and optimization of thermal pretreatment conditions. Bioresour Technol 2011; 102:2745–50.
- [17] Lay JJ, Lee YJ, Noike T. Feasibility of biological hydrogen production from organic fraction of municipal solid waste. Water Res 1999;33:2579–86.
- [18] Kim DH, Kim SH, Ko IB, Lee CY, Shin HS. Start-up strategy for continuous fermentative hydrogen production: early switchover from batch to continuous operation. Int J Hydrogen Energy 2008;33:1532–41.
- [19] Fang HHP, Liu H. Effect of pH on hydrogen production from glucose by a mixed culture. Bioresour Technol 2002;82:87–93.
- [20] Liwarska-Bizukoje E, Ledakowicz S. RNA assay as a method of viable biomass determination in the organic fraction of municipal solid waste suspension. Biotechnol Lett 2004;23: 1057–60.
- [21] Mobi T. Seaweed polysaccharides. Adv Carbohydr Chem 1953;8:315–50.
- [22] Troiano RA, Wise DL, Augenstein DC, Kispert RG, Cooney CL. Fuel gas production by anaerobic digestion of kelp. Resource Recovery and Conservation 1976;2:171–6.
- [23] Arooj MF, Han SK, Kim SH, Kim DH, Shin HS. Continuous biohydrogen production in a CSTR using starch as a substrate. Int J Hydrogen Energy 2008;33:3289–94.
- [24] Wang Y, Wang H, Feng X, Wang X, Huang J. Biohydrogen production from cornstalk wastes by anaerobic fermentation with activated sludge. Int J Hydrogen Energy 2010;35:3092–9.
- [25] Fan Y, Zhang G, Guo X, Xing Y, Fan M. Biohydrogenproduction from beer lees biomass by cow dung compost. Biomass and Bioenergy 2006;30:493–6.

- [26] Cui M, Yuan Z, Zhi X, Shen J. Optimization of biohydrogen production from beer lees using anaerobic mixed bacteria. Int J Hydrogen Energy 2009;34:7971–8.
- [27] Taguchi F, Yamada K, Hasegawa K, Taki-Saito T, Hara K. Continuous hydrogen production by Clostridium sp. strain no. 2 from cellulose hydrolysate in an aqueous two-phase system. J Ferment Bioeng 1996;81:178–80.
- [28] Ntaikou I, Gavala HN, Kornaros M, Lyberatos G. Hydrogen production from sugars and sweet sorghum biomass using Ruminococcus albus. Int J Hydrogen Energy 2008;33: 1153-63.
- [29] Nguyen T, Han S, Kim J, Kim M, Oh Y, Sim S. Hydrogen production by the hyperthermophilic eubacterium,

Thermotoga neapolitana, using cellulose pretreated by ionic liquid. Int J Hydrogen Energy 2008;33:5161–8.

- [30] Akutsu Y, Lee D, Li Y, Noike T. Hydrogen production potentials and fermentative characteristics of various substrates with different heat-pretreated natural microflora. Int J Hydrogen Energy 2009;34:5365–72.
- [31] Youn JH, Shin HS. Comparative performance between temperature-phased and conventional mesophilic two-phased processed in terms of anaerobically produced bioenergy from food waste. Waste Manage Res 2005;23:32–8.
- [32] Kim DH. Innovative two-stage fermentation system converting organic solid waste to hydrogen and methane. Doctoral thesis. Korea Inst Sci Technol; 2008.